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Atomic Structure Quiz Answers 1. All elements are composed of atoms compounds neutrons molecules 2. What is the mass number of carbon (C)? 12 6 13 C 3. How many electrons are in an atom of carbon (C)? (Remember APE! Atomic Number = # of Protons and # of... 4. How many electrons are in an atom of ...

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Atomic Structure- describe the characteristics of protons, neutrons and electrons in terms of location, charge and mass.- illustrate the

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structure of the atom by using the Bohr model, including the charge, relativemass and location of the sub-atomic particles.- use atomic mass, atomic number, and charge to identify neutral atoms, ions, and isotopes.- analyze the structure of the atom and describe the characteristics of the particles found there.- describe electron configurations for the ...

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Q. A tiny but very dense, positively charged portion of the atom that holds most of the atomic mass

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The electron dot structure depends on the number of valence electrons. To answer the question, you need to know the electron configuration of the atoms to see which one has 7 unbonded electrons, like chlorine. Fluorine, element number 9, has 2 electrons in the s sublevel (K shell). The L shell is incompletely filled, with 7 electrons.

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Atomic structure. Atoms are made from protons, neutrons and electrons. The periodic table arranges elements according to their atomic size and other properties. Part of. Chemistry.

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