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19. ChemQuest 6. Name: _____ Date:
_____ Hour: _____ Information: Specific
Heat Constants Depend on the State of
Matter . Recall that specific heat is the energy
necessary to raise the temperature of one
gram of a substance. by one degree Celsius.
The specific heat of liquid water is a constant
equal to $4.18 \text{ J/(g} \cdot \text{ }^\circ\text{C)}$. This. means that it

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takes 4.18 Joules (J) of heat energy to raise the ...

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Consider the following examples of formulas for ionic compounds: • One Na^+ (sodium ion) and one Cl^- (chloride ion) bond to ...

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ChemQuest 10 Name: _____ Date: _____
Hour: _____ Information: Bohr ' s Solar ...

11. The energy difference between the 1st
and 2nd energy levels in a certain atom is
 5.01×10^{-19} J. What frequency of light is
necessary to excite an electron in the 1st
energy level? 12. a) What is the frequency of
light given off by an electron that loses $4.05 \times$
 10^{-19} J of energy as it moves from the 2nd to
the ...

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reasoning~~

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attract and bond to each other forming an
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ChemQuest 8: Answer Key An ion with a negative charge. How is an anion formed? An anion is formed by adding electrons to an atom. How is an anion named? An anion is named by replacing the suffix of the element name with “ -ide ” followed by the word ion. What elements on the period table tend not to form ions? Why? Noble

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gases. CHEMQUEST 18 FORMING IONS

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key ionic bonding is when two ions of

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Hour: _____ Information: Relating Electron

Configurations to the Periodic Table In this

section you will see how the periodic table

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serves as a road map for writing electron configurations. Get your periodic table out and get ready. Remember that a row on the periodic table

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